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Specification and Drawings, as originally filed, with Application for Patent Serial No:
CA 2413834, on December 10, 2002, by 1017975 ALBERTA LIMITED, assignee of
Randy Miller and Christie Woodruff, for "Nitric Oxide Gas Generator".


Agent certificateur/Certifying Officer

October 31, 2006
Date

Canada

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ABSTRACT OF THE DISCLOSURE

A nitric oxide gas generator which includes a body having a dilution inlet chambers, a chemical mixing chamber, and a dilution outlet chamber. A dilution inlet for diluent gases is provided into the dilution inlet chamber. An inlet is provided to permit entry of the diluent gases into the chemical mixing chamber. An outlet is provided to permit the exit of diluted nitric oxide gas from the chemical mixing chamber to the dilution outlet chamber. A dilution outlet is provided for removal of diluted nitric oxide gas from the dilution outlet chamber. Supports are provided for supporting chemicals to be reacted to produce nitric oxide gas. A heat source is provided to heat the chemical mixing chamber in which chemicals are mixed to initiate a chemical reaction that produces nitric oxide gas.

TITLE OF THE INVENTION:

Nitric Oxide Gas Generator

FIELD OF THE INVENTION

5 The present invention relates to a nitric oxide gas generator

BACKGROUND OF THE INVENTION

10 James D. Ray and Richard A. Ogg Jr. of the Department of Chemistry of Stanford University, developed and published a method of preparing nitric oxide in 1956.

SUMMARY OF THE INVENTION

15 What is required is a nitric oxide generator which is capable of producing nitric oxide in accordance with the teachings of the method of Ray and Ogg Jr., and diluting the pure nitric oxide into concentrations that have utility.

20 According to the present invention there is provided a nitric oxide gas generator which includes a body having a dilution inlet chambers, a chemical mixing chamber, and a dilution outlet chamber. A dilution inlet for diluent gases is provided into the dilution inlet chamber. An inlet is provided to permit entry of the diluent gases into the
25 chemical mixing chamber. An outlet is provided to permit the exit of diluted nitric oxide gas from the chemical mixing chamber to the dilution outlet chamber. A dilution outlet is provided for removal of diluted nitric oxide gas from the dilution outlet chamber. Supports are provided for
30 supporting chemicals to be reacted to produce nitric oxide gas. A heat source is provided to heat the chemical mixing chamber in which chemicals are mixed to initiate a chemical reaction that produces nitric oxide gas.

BRIEF DESCRIPTION OF THE DRAWINGS

These and other features of the invention will become more apparent from the following description in which reference is made to the appended drawings, the drawings are
 5 for the purpose of illustration only and are not intended to in any way limit the scope of the invention to the particular embodiment or embodiments shown, wherein:

FIGURE 1 is a side elevation view, in section, of a nitric oxide generator constructed in accordance with the
 10 teachings of the present invention.

FIGURE 2 is a process flow diagram which demonstrates the utility of the nitric oxide generator illustrated in **FIGURE 1**.

DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENT

The preferred embodiment, a nitric oxide gas generator will now be described with reference to **FIGURES 1** and **2**.

DESCRIPTION OF NITRIC OXIDE GAS GENERATOR

Existing Patent:

(Contribution from the Department of Chemistry, Stanford University)
 A New Method of Preparing Nitric Oxide
 By James D. Ray and Richard A. Ogg, Jr.
 Received July 25, 1956

25 A new method of preparing nitric oxide which involves heating to a temperature slightly above 300 degrees a dry powdered mixture of potassium nitrite and nitrate, chromic oxide and ferric oxide has been perfected. The nitric oxide so produced contained only a fraction of a per cent. of impurity. (see attached photocopy 5993 for complete description)

1. Description of problem

One of the problems with the above method is the lack of accurate temperature control.

Solution

30 In order to produce reliable quantity of nitric oxide gas the temperature must be accurately controlled we shall do this by means of an electronically controlled electric heater and compressing the chemical mixture into a lifesaver shape, which will allow consistent repeatable heat transfer from the heater to the mixture.

2. Description of problem

It is necessary to capture the nitric oxide gas as it is produced and shield it from moisture, air and other unwanted contaminants.

Solution

35 An integral heater and gas capture vessel (turtle shell) with appropriate fittings will resolve the problem. See drawing 1.

3. Description of problem

Chemical mixture inconsistency will have an adverse effect on the purity, quality, quantity and repeatability of the generated gas. Inconsistency of the generated gas is caused by settling out of the mixture components due to variations in temperature, vibration and other mechanical means.

Solution

- 5 In order to resolve inconsistency in the chemical mixture the chemicals will be calcined at 950 degrees Celsius in order to remove the water of hydration and then adequately mixed and compressed into a lifesaver configuration. This will prevent separation of the chemical mixture during transportation, generation of gas, shipping and handling.

4. Description of problem

The chemical mixture will only produce pure nitric oxide (one million parts per million of nitric oxide gas is generated). What is needed is a method of varying the concentration of nitric oxide gas.

Solution

- 10 Dilution of pure nitric oxide is achieved by the entrainment of air, nitrogen, oxygen, other inert gases or any combination thereof into the integral captured gas container. See diagram 2.

5. Description of problem

Impurities in the final product due to potassium nitrite not being of sufficient purity (contains about 10% potassium nitrate) are unacceptable.

Solution

- 15 Obtain commercially available potassium nitrite (potassium nitrite was not available as an article of commerce in 1956).

6. Description of problem

The process of creating nitric oxide as described in the 1956 method of preparation is impractical for transportation.

Solution

Construct self contain generator (turtle). See drawing 3.

7. Description of problem

Control of the total amount of gas produced and the rate of production.

Solution

Introduction of non-reactant binding reagents, configuration (lifesaver), incremental increase of reagents of known volume (size and number of lifesavers, automatic timer for heater).

8. Description of problem

Lack of shock resistance, lack of stability of reagents to physical abuse

Solution

Generator enclosed in turtle shell and lifesaver configuration

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ELEMENTS OF THE NITRIC OXIDE GAS GENERATOR

Element 10 - A Chemical Mixture

Description - The mixture is 63.75 g. (0.750 mole) potassium nitrite, 25.25 g. (0.250 mole) potassium nitrate, 76 g. (0.50 mole) chromic oxide and 120 g. (0.752 mole) ferric oxide.

5

Element 11 - A Cartridge Heater

Description - A commercial heater capable of 310 degrees Celsius ^{with temperature} control device

Element 12 - Integral Gas Capture Device (Turtle)

Description - A container that captures gas created when the chemical mixture is heated.

Element 13 - A Chemical Mixture Configuration

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Description - A chemical mixture is compressed into a lifesaver shape that allows convenient placement of the compressed chemical onto the heater probe.

Element 14 - Power Source

Description - Commercial heater powered by an electrical outlet or a rechargeable battery (depends on the volume of gas required and the portability required).

Element 15 - Plumbing and Fittings

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Description - Commercial plumbing and fittings used as required to direct gas to the desired location.

Element 16 - Gas Dilution Device

Description - This device can be a compressed nitrogen cylinder or an air entrainment device such as a pump with a calibrated orifice.

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In this patent document, the word "comprising" is used in its non-limiting sense to mean that items following the word are included, but items not specifically mentioned are not excluded. A reference to an element by the indefinite article "a" does not exclude the possibility that more than one of the element is present, unless the context clearly requires that there be one and only one of the elements.

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It will be apparent to one skilled in the art that modifications may be made to the illustrated embodiment without departing from the spirit and scope of the invention as hereinafter defined in the Claims.

**THE EMBODIMENTS OF THE INVENTION IN WHICH AN EXCLUSIVE
PROPERTY OR PRIVILEGE IS CLAIMED ARE DEFINED AS FOLLOWS:**

- 5 1. A nitric oxide gas generator, comprising:
- 10 a body having a dilution inlet chambers, a chemical mixing chamber, and a dilution outlet chamber, a dilution inlet for diluent gases being provided into the dilution inlet chamber, an inlet being provided to permit entry of the diluent gases into the chemical mixing chamber, an outlet being provided to permit the exit of diluted nitric oxide gas from the chemical mixing chamber to the dilution outlet chamber, a dilution outlet being provided for removal of diluted nitric oxide gas from the dilution outlet chamber;
- 15 supports for supporting chemicals to be reacted to produce nitric oxide gas; and
- a heat source to heat the chemical mixing chamber in which chemicals are mixed to initiate a chemical reaction that produces nitric oxide gas.

ELEMENTS OF NITRIC OXIDE GAS GENERATOR

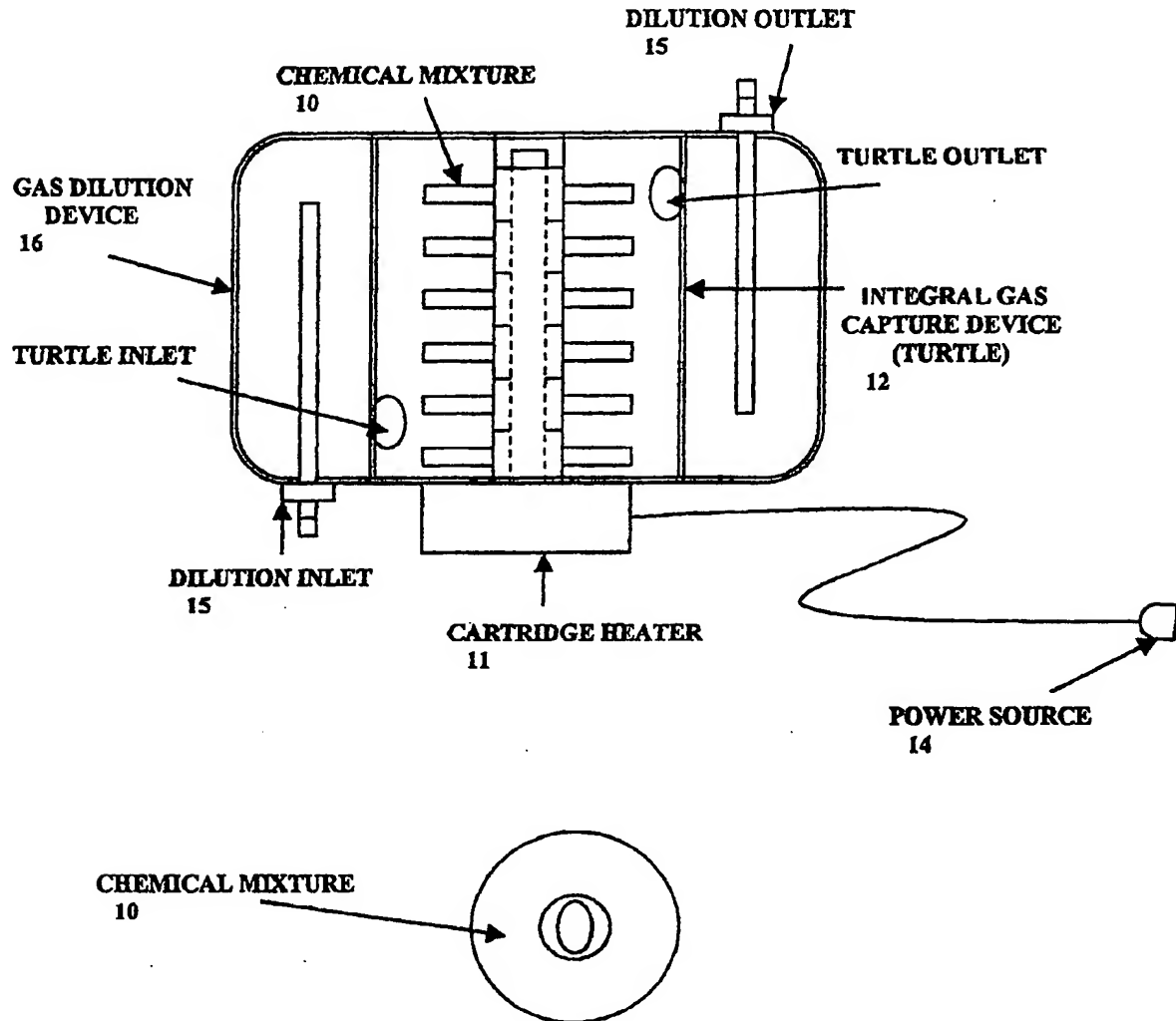
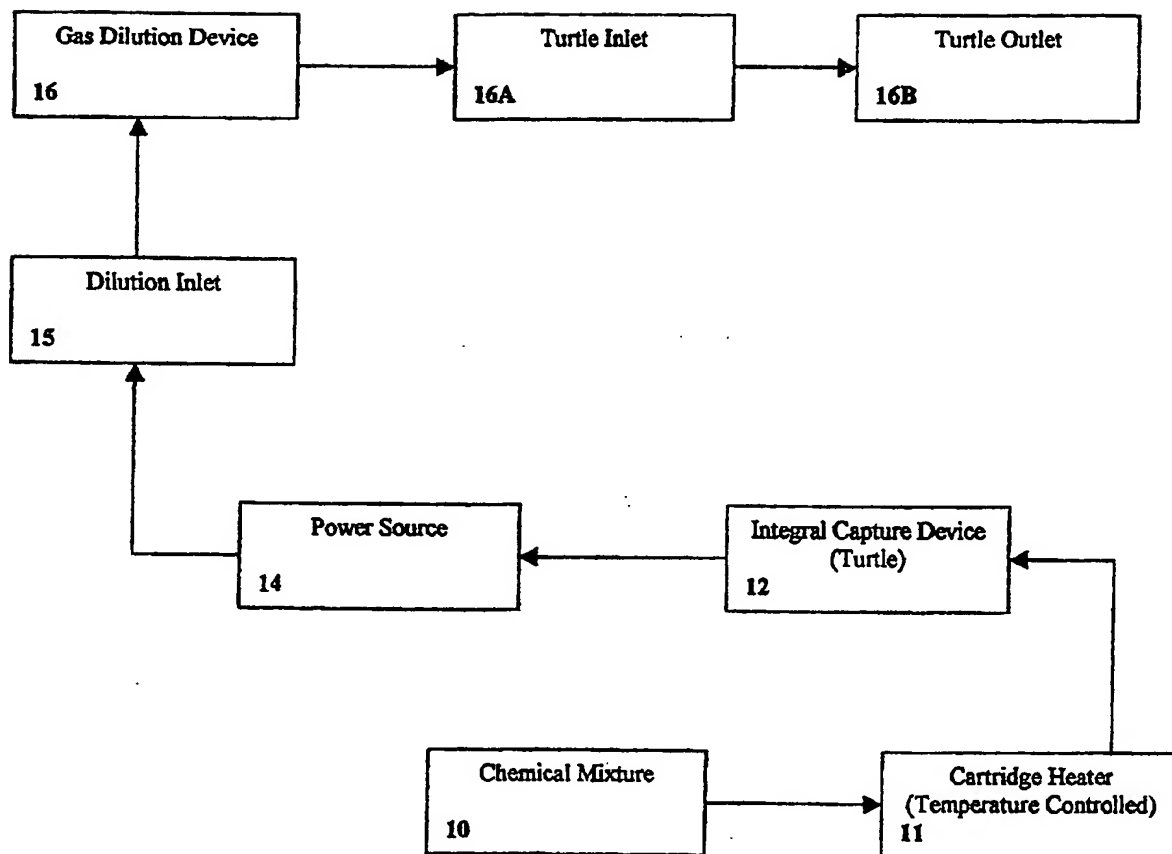


FIG. 1

ELEMENTS OF NITRIC OXIDE GAS GENERATOR (PROCESS FLOW DIAGRAM)

**FIG. 2**